MODELING CO₂ MICROSOLVATION: MICROWAVE SPECTROSCOPIC STUDIES OF DIFLUOROETHYLENE (DFE)/CO₂ CLUSTERS, (DFE)₁(CO₂)_x, FOR A TRIMER, TETRAMER, AND PENTAMER

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Microwave spectroscopy allows for analysis of weakly-bound clusters in a mixture of difluoroethylene (DFE) and CO_2 . The present study probes variations in interactions and orientations of DFE and CO_2 within weakly-bound clusters as cluster size increases. Four chirped-pulse FTMW spectra of DFE/ CO_2 mixtures were obtained from 2-8 GHz, where the concentration of CO_2 was varied from 1% to 4%, with a constant DFE concentration of 1%. This experimental design allowed variation in intensity to be observed based on the variation of CO_2 concentration, where the pattern of intensity variation was used to identify transitions belonging to a particular cluster. In addition, patterns of intensity variation provided information about the size and DFE: CO_2 ratio of the cluster. Using these methods based on intensity variation analysis, a three separate sets of transitions, each with unique intensity variation patterns, were extracted from the original raw spectra. Cluster composition was hypothesized based on further evidence from the intensity variation analyses, leading to compositions of $(\mathrm{DFE})_1(\mathrm{CO}_2)_2$, $(\mathrm{DFE})_1(\mathrm{CO}_2)_3$, and $(\mathrm{DFE})_1(\mathrm{CO}_2)_4$. Fitted rotational constants for the spectra were compared to the results of ab initio calculations, which further supported hypothesized cluster compositions for the trimer, tetramer and pentamer. These results indicate that instead of forming a solvation shell around DFE, CO_2 molecules appear preferably to interact with other CO_2 molecules to form arrangements more closely resembling pure CO_2 clusters, with DFE on the outside of the cluster.

^aH. Fino, R.A. Peebles, S.A. Peebles, C. West, B. Pate, International Symposium on Molecular Spectroscopy (Virtual), Talk FH12, June 25, 2021; R.A. Peebles, S.A. Peebles, P. Kannangara, H. Fino, International Symposium on Molecular Spectroscopy (Virtual), Talk FH13, June 25, 2021