

## LINE INTENSITY MEASUREMENTS AND FAR-WING INTENSITY REDISTRIBUTION IN THE 0.76 $\mu\text{m}$ O<sub>2</sub> BAND

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Molecular oxygen (O<sub>2</sub>) has a well-known and uniform molar fraction within Earth's atmosphere. Because of this, the near-infrared O<sub>2</sub> molecular absorption bands, centered at 0.76  $\mu\text{m}$  and 1.27  $\mu\text{m}$ , are commonly used in remote sensing (TC-CON, COCCON) and satellite missions (GOSAT, OCO-2/3, SCIAMACHY) to measure atmospheric air mass. For these missions, physics-based spectroscopic models and experimentally and theoretically determined line-by-line parameters are used to predict the temperature- and pressure-dependence of the absorption cross-section as a function of wave number, pressure, temperature, and water vapor concentration. The accuracy of atmospheric retrievals requires the spectroscopic model to describe all relevant physics. The Hartmann-Tran line profile is recommended for high-resolution spectroscopy and can be reduced to the well-known Voigt profile [1]. This profile assumes that collisions occur instantaneously. However, when this assumption is not satisfied, the finite duration of collisions leads to a redistribution of the line intensity from the line cores to the far wings, an effect that increases with pressure [2-4]. Failure to account for this redistribution leads to an apparent depletion in the core line intensity as a function of pressure, such that retrieved concentration would yield a bias if determined from absorption cross-section based on observations acquired at substantially different pressures. This work reports cavity ring-down spectroscopy measurements of line intensities and pressure-dependent intensity redistribution in the 0.76  $\mu\text{m}$  O<sub>2</sub> band. The core intensity depletion magnitude and rotational quantum number dependence are compared to those calculated by renormalized classical molecular dynamic simulations. Additionally, we discuss how including the previously unaccounted-for physics might affect satellite and remote sensing retrievals.

[1] Tennyson, J., et al., *Pure and Applied Chemistry*, 2014. 86(12): p. 1931-1943. [2] Reed, Z.D., et al., *Phys Rev Lett*, 2023. 130(14): p. 143001. [3] Tran, H., et al., *J Chem Phys*, 2023. 158(18). [4] Tran, H., et al., *Phys Chem Chem Phys*, 2023. 25(15): p. 10343-10352.